

## REMOVING METHOD FOR CHROMIUM AND NICKEL CONTAINED IN IRON CHLORIDE AQUEOUS SOLUTION

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### Abstract of **JP62192588**

**PURPOSE:** To regenerate an FeCl<sub>2</sub> aq. soln. to an FeCl<sub>3</sub> aq. soln. by adding an iron piece to an FeCl<sub>3</sub> waste liquid contg. Cr and Ni which is obtained by treating stainless steel and precipitating and removing Cr and thereafter by adding iron powder to the filtered liquid to precipitate and remove Ni and blowing Cl<sub>2</sub> into the residual liquid contg. FeCl<sub>2</sub>.

**CONSTITUTION:** Since Cr and Ni are contained in a liquid wherein stainless steel is subjected to etching treatment by an FeCl<sub>3</sub> aq. soln., an iron piece of the excess amount than the amount necessary to change the residual FeCl<sub>3</sub> to FeCl<sub>2</sub> is added and allowed to react with each other at 1-3 pH at 50-90 deg.C temp. to precipitate Cr as Cr(OH)<sub>3</sub> and thereafter it is filtered, removed and recovered. Iron powder having  $\geq 150$  mesh is added to the FeCl<sub>2</sub> aq. soln. of the residual liquid in  $\geq 1$  time mol for contained Ni and allowed to react at 80 deg.C for 2-8hr to precipitate the contained Ni and it is filtered and recovered. The FeCl<sub>2</sub> aq. soln. wherein Cr and Ni are removed therefrom is blown with gaseous Cl<sub>2</sub> and regenerated as the FeCl<sub>3</sub> aq. soln. and reutilized as a treating liquid of stainless steel.

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